

*Webinar*  
**Analysis of  
Fresh Water with  
Auto Titrators**



# House Keeping



60 minute  
webinar



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# Our Presenters



**Stefan Kaus**

*Titration Product Manager*

- More than 30 years experience in titration application, automation and development of titrators



**Dr. Jens Hillerich**

*Head of Titration Applications  
PhD in Engineering*

- 5 years of experience in titration applications, automation and development of titrators



**Dr. Tao Su**

*Product Manager &  
PhD in Environmental Sciences*

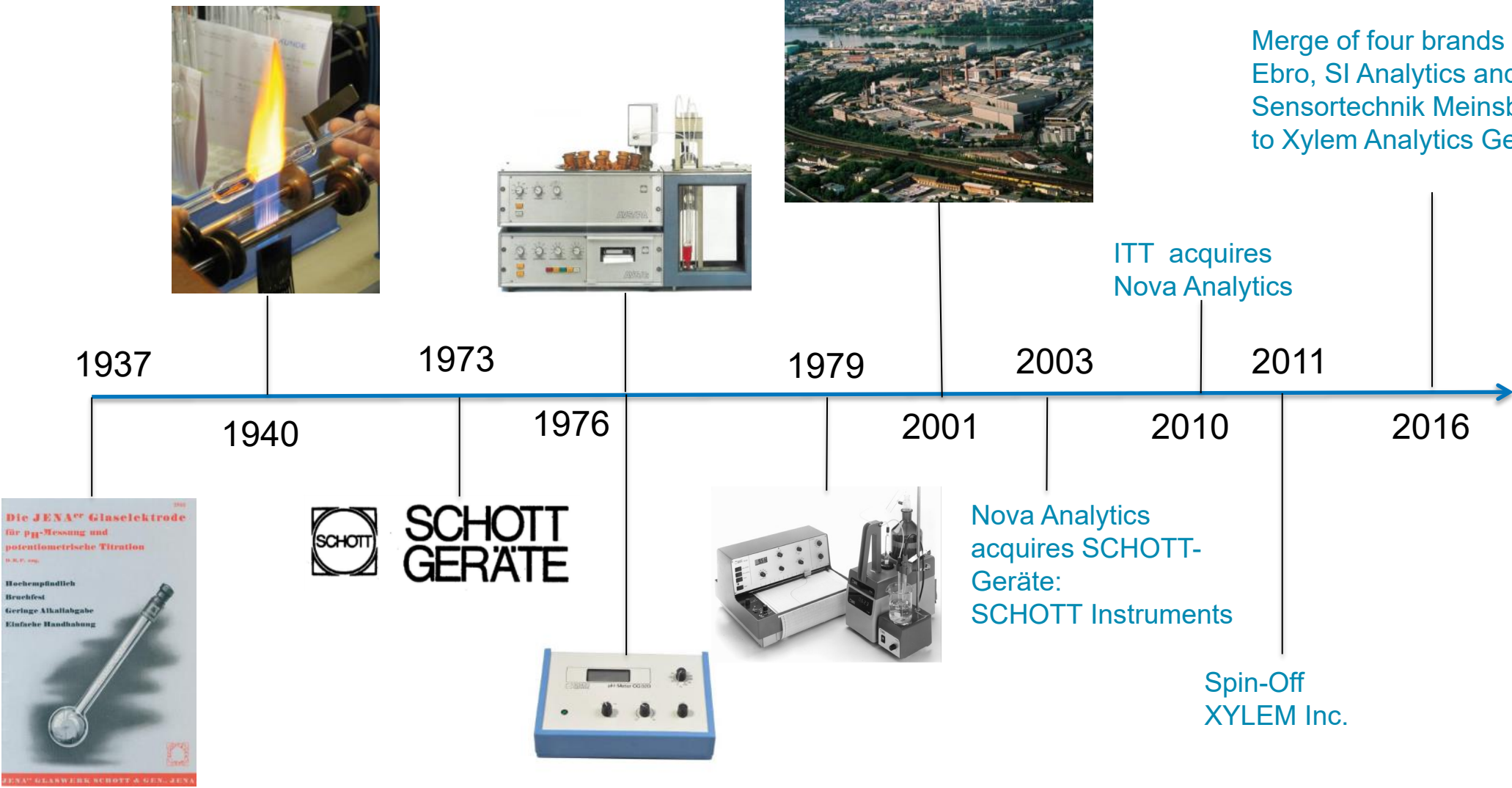
- Responsible for product sales and promotion of WTW online products throughout North Asia with 3 years with Xylem (Analytics)

# Poll Question #1



Do you have an automatic titration system in your laboratory or workplace?

# Evolution of SI Analytics



# SI Analytics Head Office

- Located in Mainz (South West Germany)
- 30 km to Frankfurt (Rhein-Main) Airport



# Xylem Analytics - SI Analytics®

- Site is on premises of Schott AG
- Production, R&D, order processing, sales and marketing
- 120 employees



# SI Analytics® Product Overview

Electrodes and meters for lab, field and process



Piston burettes and automatic titrators



Capillary Viscometry & Automatic Viscosity Systems





# Why is Titration still used?

- Well known method  
The method is old (about 200 years!) and well known in the lab.
- Versatile  
Hundreds of methods exist for all ranges for contents from ppm up to the 100 % range.
- Cost-effective  
With a simple equipment you can already carry out a titration.  
Low purchase costs and low follow-up costs
- Fast and accurate  
The titration time is usually between 1-3 min. The reproducibility is often 0.1 % or even better



# What is Titration?

- **Absolute method**

The result is calculated directly from the consumption of the titrant.  
Cumbersome calibration is not necessary.

- **Quantitative analytical method**

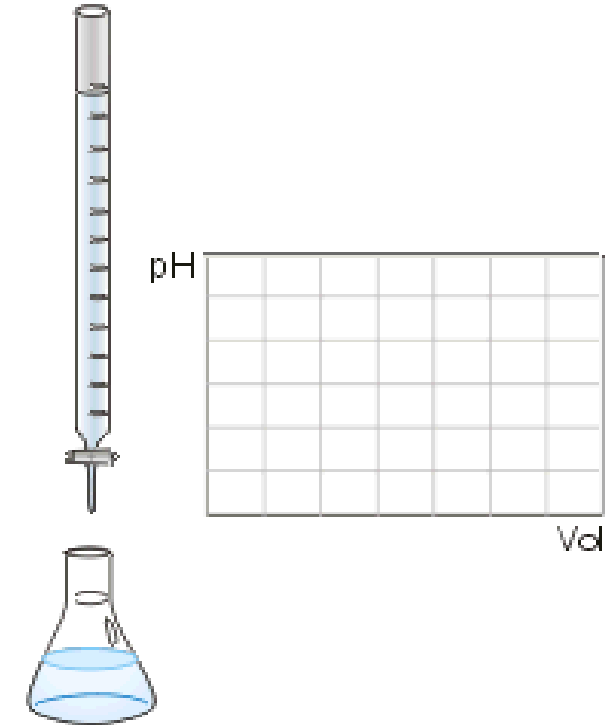
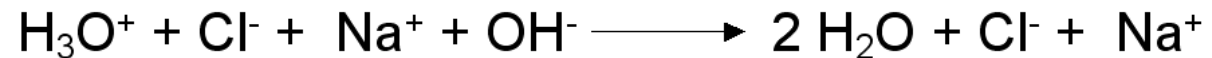
This means that you want to know how much is contained in a sample and not which substance. The substance is (mostly) known.

- **General principle**

Titration is the addition of a liquid reagent of known concentration (titrant) to a sample of unknown concentration.

# What is Titration?

- The sample is dissolved in water or another suitable solvent, such as ethanol, toluene, etc.
- A chemical reaction takes place during the addition of the titrant. This reaction is known, clear and always takes place in the same way.
- The end of the reaction is indicated by either a color indicator or a potential change (mV, pH,  $\mu\text{A}$ ...).



## Poll Question #2



Which industry are you working in?

# Where is Titration used?

## Chemical/Pharma industry

- Process control
- Product and raw material quality control
- All type of potentiometric titration and
- Water determination according to KF

## Environmental/water labs

- Water hardness
- pH and Alkalinity, EC
- Chlorine and chloride
- COD
- And many

## Food industry

- Water determination according to KF
- Total acidity in beverages
- Stability of fats and oils
- Water quality (hardness,

## Electroplating

- Acid and bases
- Metals
- Surfactants

## Petrochemistry

- Water determination according to KF
- TAN and TBN
- Chloride
- H<sub>2</sub>S and mercaptanes

# Main Applications in Fresh Water Analysis

- pH value: APHA 4500-H<sup>+</sup>/DIN EN ISO 10523/ASTM D1293/EPA 150.1
- Electrical Conductivity ISO 7888/EN 27888/ASTM D1125/EPA 120.1
- Alkalinity: APHA 2320/DIN EN ISO 9963-1/ASTM D1067
- Calcium/Magnesium and total water hardness: ASTM D511, D1126/ISO 6058, ISO 6059, EPA 130.2
- Chlorine: ASTM 1253, EPA 330-1 and 330-2, DIN EN ISO 7393-1 and 7393-3
- Chloride: ASTM D 512, APHA 4500-Cl/D
- Fluoride: ASTM D 1179, EPA 340.2, ISO 10359-1
- Ammonium: ISO 5664, EPA 359.2, EPA 305.3, ASTM D1426

# SI Analytics Titrators & Autosamplers



# Sample Changer TW 7200 – the flexible solution

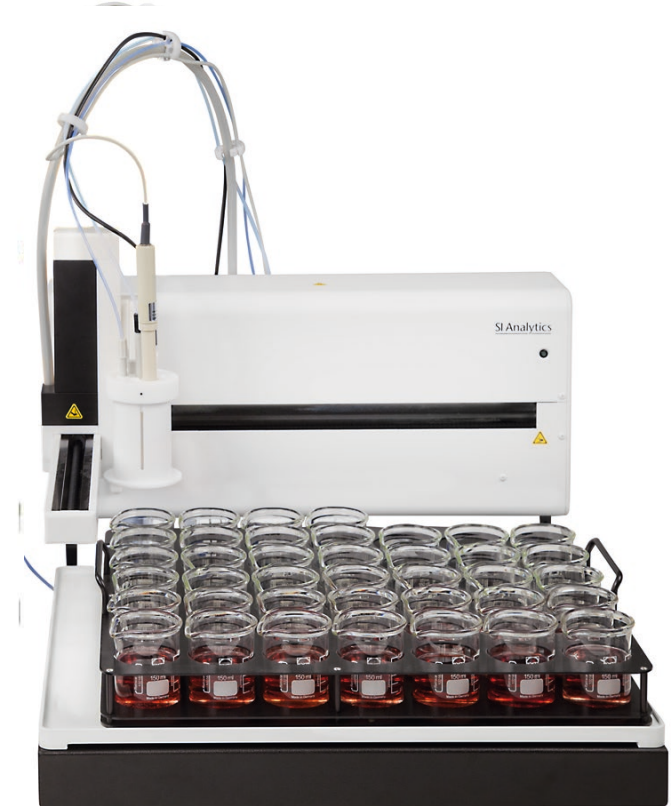
- Various titration heads and sample trays from 12 up to 38 positions available for titration application.
- The beaker size varies from 50 ml up to 600 ml.
- A sample tray for COD titration vessels is also available for the TW 7200





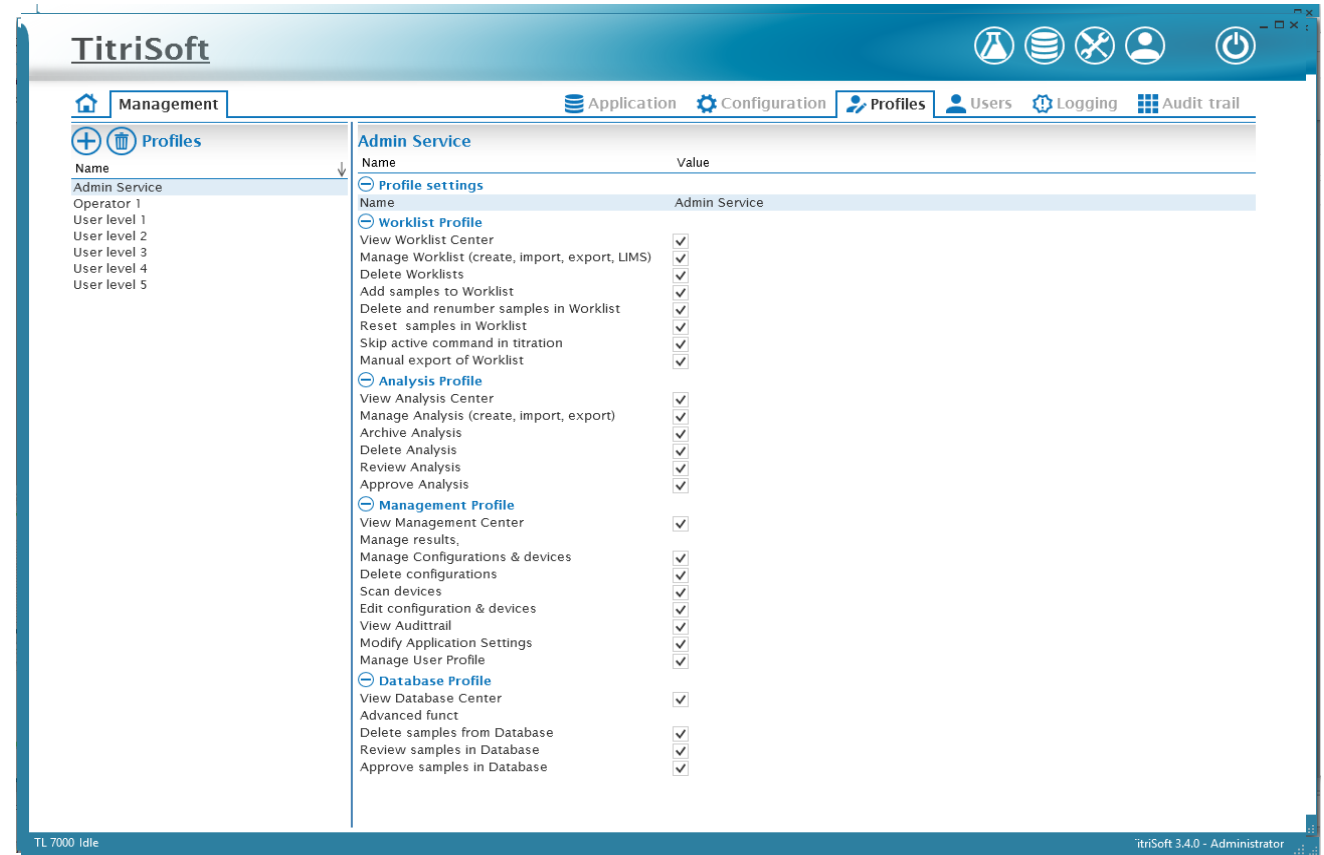
# Sample Changer TW 7450 – for high sample throughput

- Various titration heads and sample racks 2 with 42, 48 and 72 positions.
- The beaker size varies from 50 ml up to 250 ml.



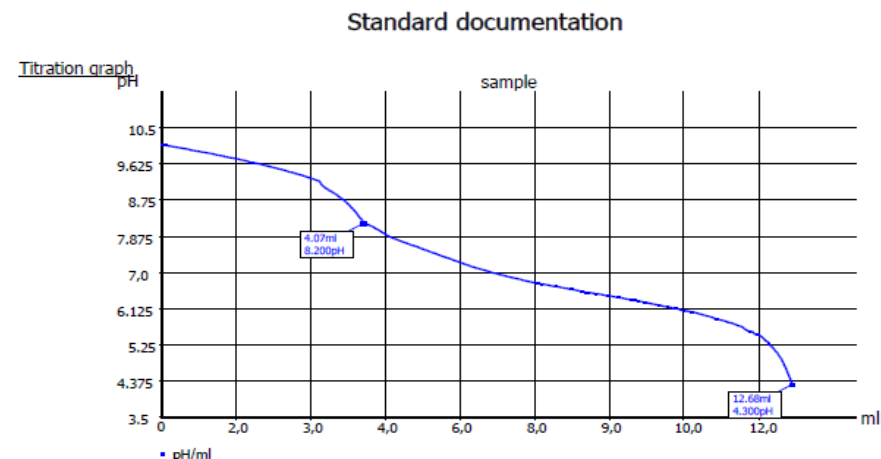
# PC Software TitrSoft 3.5 & 3.5P

- The PC software for the TitroLine<sup>®</sup> 7000 and further titrators, piston burettes and autosampler
- With TitrSoft you have the full control of each part of the method. All settings are made with the software



# Alkalinity

- Determination of the carbonate and hydrogen carbonate hardness of water
- The water sample is titrated with HCl (0.01 – 0.1 mol/l) to two fixed pH endpoints 8.3 and 4.5 or inflection points



# pH, Conductivity and Alkalinity (automation)

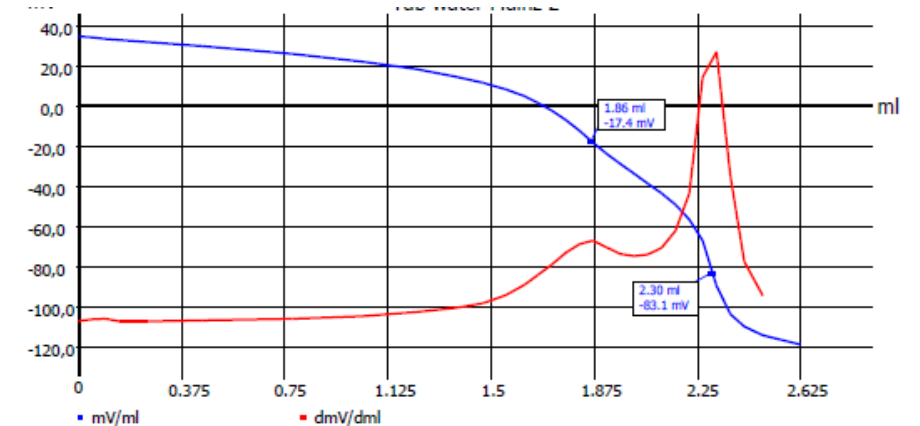
- With the TitroLine® 7800 you can measure all three parameters in one beaker one after another
- Public water labs with many sample are using autosamplers
- Preferred the x/y TW 7450-42 or 48 are used for this application but also the TW 7450-72 can be used
- First the EC, temperature and the pH is measured. Then the alkalinity is titrated with HCl

The screenshot displays the SI Analytics software interface. At the top, there is a header bar with the title 'SI Analytics' and several icons. Below the header, a navigation bar shows 'TL 7800 - TL7000-TW7400', 'CSB - Permanganatindex', 'Basenkapazität 1', and 'Worklist Analysis Curve'. The main area is divided into two panels. The left panel, titled 'Worklists', shows a table with columns 'Name' and 'Modified'. The right panel, titled 'Worklist: pH + EL + Ks neu', shows a detailed table with columns: 'Nr', 'Position', 'Status', 'Method', 'Sample ID', 'Density', 'Sample amount', 'Date', 'Temp. °C', 'pH', 'EL µS/cm', 'ml pH 8.2', 'P-value mmol/l', 'ml pH 4.3', and 'm-value mmol/l'. The table contains 8 rows of data, with the first three rows marked as 'Ready' and the last three as 'Pending'.

Nr	Position	Status	Method	Sample ID	Density	Sample amount	Date	Temp. °C	pH	EL µS/cm	ml pH 8.2	P-value mmol/l	ml pH 4.3	m-value mmol/l
1	3	Ready	pH + EL	6836			13.04.2016 12:22	23,2	7,34	656				
2	4	Ready	pH + EL	6840			13.04.2016 12:23	23,3	9,00	154				
3	5	Ready	pH + EL	6906			13.04.2016 12:24	23,4	7,72	1273				
4	6	Ready	pH + LF + Ks	6933		100	13.04.2016 12:25	23,4	6,92	803	0,00	0,00	6,59	6,59
5	7	Ready	pH + LF + Ks	6934		100	13.04.2016 12:30	23,5	7,28	523	0,00	0,00	3,34	3,34
6	8	Pending	pH Checkstandard	6935	1.0034	90								
7	9	Pending	pH Checkstandard	6936	1.0045	90								
8	10	Pending	pH Checkstandard	6937	1.05	95								

# Determination of Total Hardness

- The total hardness is the sum of all alkaline earth metals (Ca, Mg...) in water
- After the addition of  $\text{NH}_4\text{OH}/\text{NH}_4\text{Cl}$ -buffer solution and an indicator the water sample is titrated with  $\text{EDTA-Na}_2$  to one inflection point
- As sensor for the colorimetric indication we use the optical sensor OptiLine 6
- Also a Ca or Cu-ISE can be used to determine the total hardness. With a Ca-ISE is it also possible to titrate the Ca and the Mg- concentration with one titration.



## Method data

Method name:	Ca and Mg	Titration duration:	5 m 10 s
End date:	23.11.12	End time:	16:11:23

## Titration data

Start mV:	34.6 mV	Pattern:	100.00000 ml
		End mV:	-118.9 mV
EQ1:	1.863 ml / -17.4 mV	Calcium:	74.7 mg/l
EQ2:	2.303 ml / -83.1 mV	Magnesium:	10.7 mg/l

# Analysis of fresh water with auto titrators

Live demo of pH + EC + Alkalinity and Total hardness  
parallel on two titration systems

## Poll Question #3



Which titration application would you like to see us cover in our next webinar?

*Webinar*  
**Analysis of  
Fresh Water with  
WTW Products**





# Dr. Tao Su



## BACKGROUND

**PhD in Environmental Sciences  
University of Tokyo**

- Xylem Product Manager
- Responsible for product sales and promotion of WTW online products throughout North Asia.
- 3 years with Xylem (Analytics)

# Topics

- Parameters to validate for fresh water and drinking water
- How WTW lab products can help you

## Poll Question #4



In your routine water quality analysis, which parameters do you usually measure by electrode and meter?

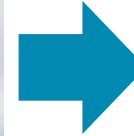
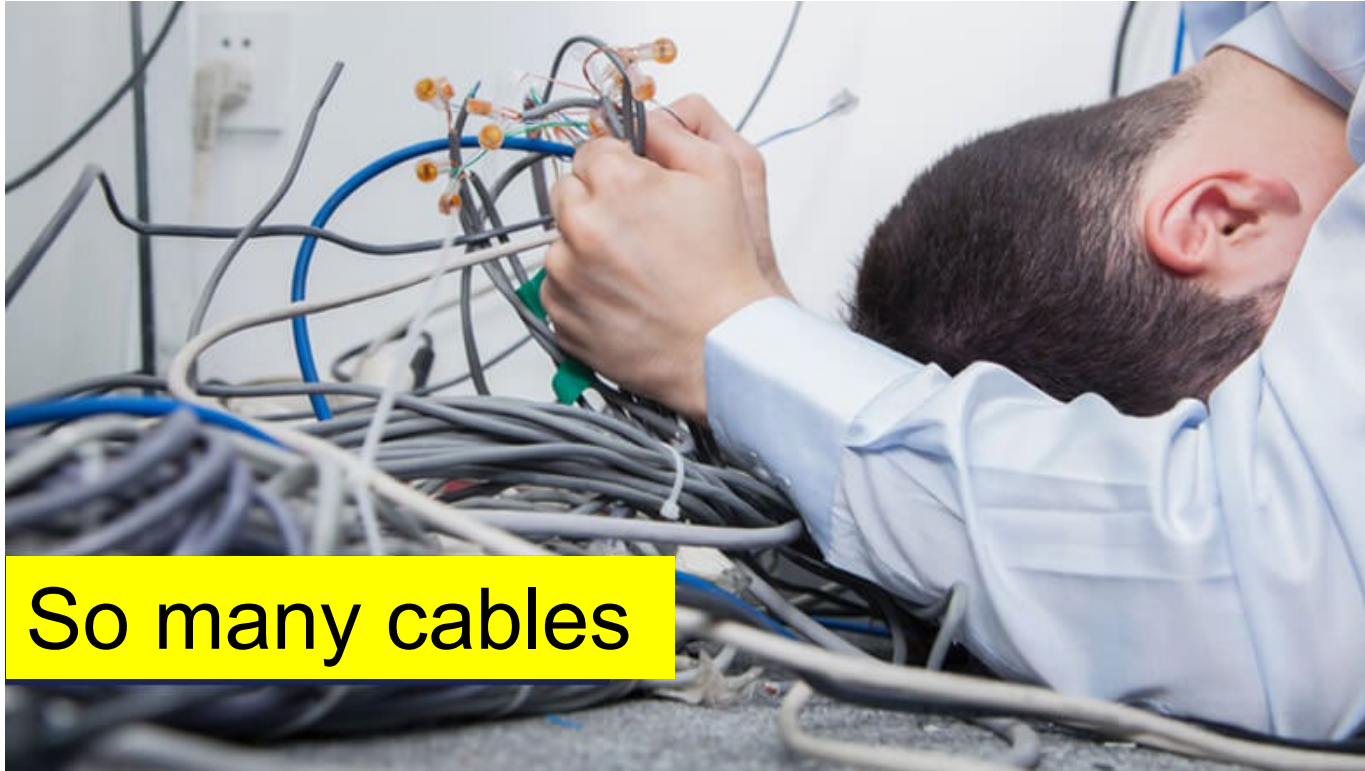
# Water Quality Parameters

	Physical parameters	Chemical parameters
1	Turbidity	pH
2	Temperature	Acidity
3	Color	Alkalinity
4	Taste and odor	Chloride
5	Solids	Chlorine residual
6	Electrical conductivity (EC)	Sulfate
7		Nitrogen
8		Fluoride
9		Iron and manganese
10		Copper and zinc
11		Hardness
12		Dissolved oxygen
13		Biochemical oxygen demand (BOD)
14		Chemical oxygen demand (COD)
15		Toxic inorganic substances
16		Toxic organic substances
17		Radioactive substances

\* Nayla Hassan Omer, Water Quality Parameters, Water Quality - Science, Assessments and Policy

# WTW – leading you into the future

Are you getting tangled in your cables?



**WTW wireless solution**



# IDS Wireless Sensor



PH

FDO

Turbidity

Cond. & TDS & Salinity



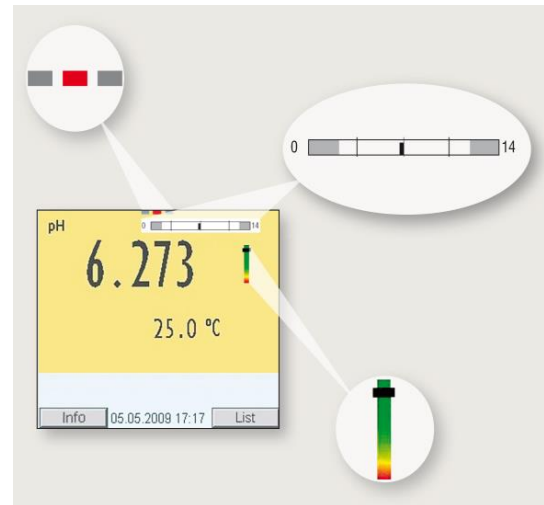
IDS WLM Kit

- **Measure everywhere:** Under laminar flow benches, laboratory hoods, glove boxes, out door (Range around 10 m)
- **Continuous operating times:** pH/ORP 60 h, conductivity 30 h, D.O. 9 h, turbidity 5 h
- **Waterproof IP 66** rated for outdoor applications

# Handheld and Benthop IDS Meters – wireless ready



## Channel display



## CMC (Continuous measurement Control)

monitoring the correct measuring range: gives indication if the current value is inside the calibration range and warning if it is outside

## QSC (Quality Sensor control)

function for IDS pH electrodes, It evaluates the condition of an IDS-pH sensor individually and with a very fine grading from 0~100 (from green to red)

## GMP/GLP

User administration with different access levels

- Three galvanically isolated inputs for fail safe signals
- Brilliant color graphic display for clear information
- 1 meter with +10 parameters
- Wireless ready

# WTW Photometer

## Photolab 7600



### Chemical parameters

pH
Acidity
Alkalinity
Chloride
Chlorine residual
Sulfate
Nitrogen
Fluoride
Iron and manganese
Copper and zinc
Hardness
Dissolved oxygen
Biochemical oxygen demand (BOD)
Chemical oxygen demand (COD)
Toxic inorganic substances
Toxic organic substances
Radioactive substances



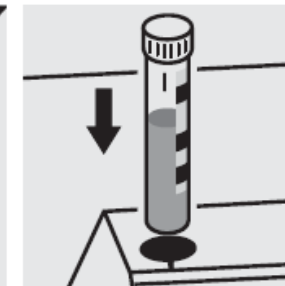
Pipette 4.0 ml of AC-1 into a round cell.



Add 1.0 ml of the sample with pipette, close the cell with the screw cap, and mix.



Add 0.50 ml of AC-2 with pipette, close the cell with the screw cap, and mix.

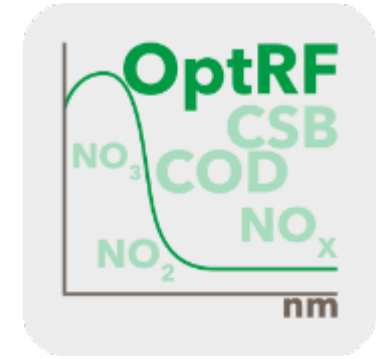


Place the cell into the cell compartment. Align the mark on the cell with that on the photometer.

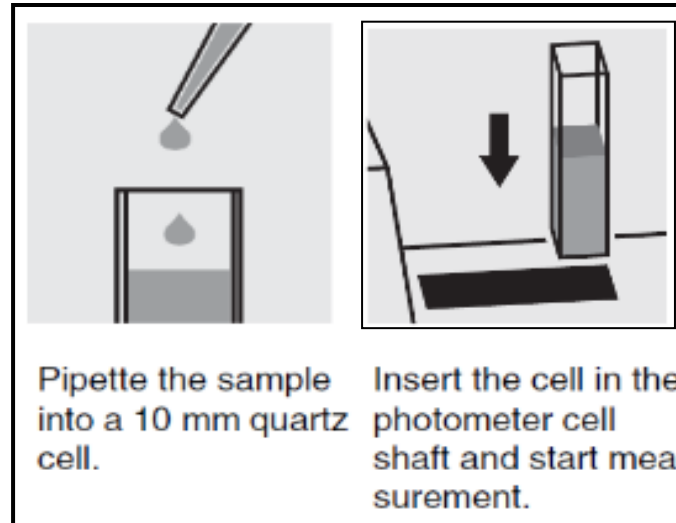


# Pioneering online procedures transferred to the Lab!

## Optical Reagent-Free Measurement OptRF COD, NO<sub>3</sub>, NO<sub>2</sub> - Just measure!



Select method (all)	16.01.15 9:52
Search: <input type="text"/>	
<b>3001 CODt_H_Outlet_10</b>	<b>COD 0.0 - 75.0 mg/l</b>
3002 CODs_H_Outlet_10	COD 0.0 - 75.0 mg/l
3003 NO3_H_Outlet_10	NO <sub>3</sub> -N 0.0 - 3.0 mg/l
3004 NO2_H_Outlet_10	NO <sub>2</sub> -N 0.00 - 4.00 mg/l
Last used	



OptRF measurement	16.01.15 9:52
[ZERO 16.01.15 9:51]	
[MQ 1]	
<b>2.3</b> mg/l	
Raw value: #2.0 mg/l	
3003:NO3_H_Outlet	NO <sub>3</sub> -N
10 mm	0.0 - 3.0 mg/l
Setup	Method list Citation form

- *Faster* than any digestion (COD standard method: 2.5 hrs.)
- *Cost free* no need of chemicals
- *Ecofriendly* no potassium Dichromate, Mercury...
- *Non-hazardous, non carcinogenic*

# Turbidity



PhotoFlex® Turb

- Photometer, pH, Turbidity, ORP



Turb® 750

- High precision measurement procedure: **IRPC**
- Fast and easy handling including quick routine

# Q&A Session

## Poll Question #5



Would you like our product specialists to contact you with more information?

# Questions?

## CONTACT US

**Stefan Kaus**  
[stefan.kaus@xylem.com](mailto:stefan.kaus@xylem.com)

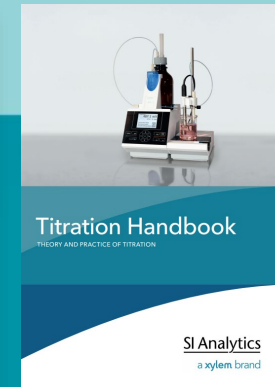
**Dr. Jens Hillerich**  
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**Dr. Tao Su**  
[tao.su@xylem.com](mailto:tao.su@xylem.com)

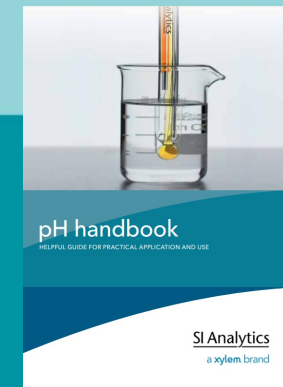
**Xylem Marketing**  
[info.apac@xylem.com](mailto:info.apac@xylem.com)



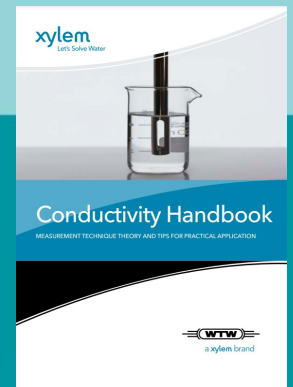
*Automated Titration vs. Manual Titration*



*Titration  
Handbook*



*pH  
Handbook*



*Conductivity  
Handbook*